

# **CO<sub>2</sub> Utilization in Concrete Production**

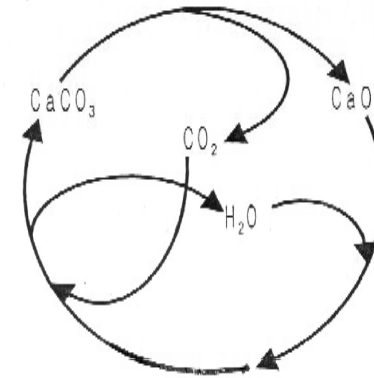
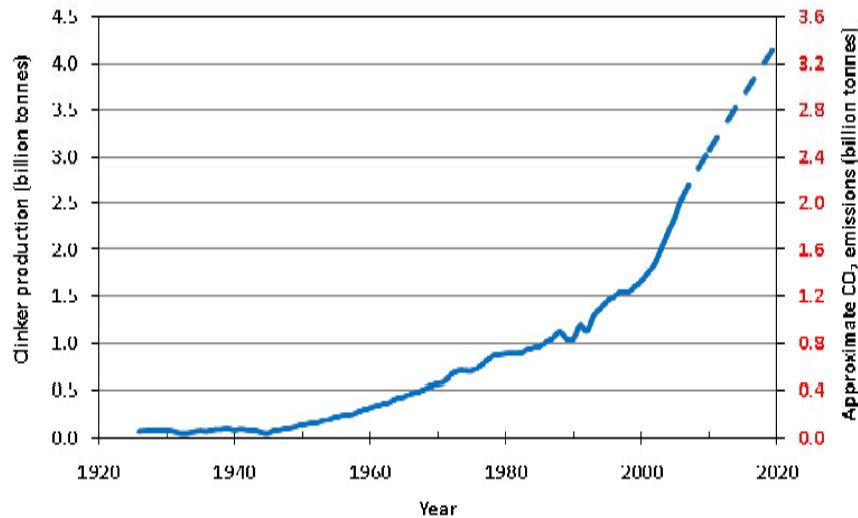
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# CO<sub>2</sub> Emission and Its Utilization



- Utilization of CO<sub>2</sub> in concrete production to reduce emission and improve performance

## Use of CO<sub>2</sub> as Curing Agent

- Weathering carbonation at late age:
  - $\text{Ca(OH)}_2 + \text{CO}_2 \rightarrow \text{CaCO}_3 + \text{H}_2\text{O}$
  - $3\text{CaO} \cdot 2\text{SiO}_2 \cdot 3\text{H}_2\text{O} + 3\text{CO}_2 \rightarrow 3\text{CaCO}_3 + 2\text{SiO}_2 \cdot 3\text{H}_2\text{O}$
  - CO<sub>2</sub> uptake is difficult to quantify
  - Passive carbonation can be detrimental
- Carbonation curing at early age (Young, 1974):
  - $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{CO}_3$
  - $\text{C}_3\text{S} + 1.2\text{H}_2\text{CO}_3 \rightarrow \text{C}_{1.4}\text{SH}_{0.6} + 1.2\text{CaCO}_3 + 0.6\text{H}_2\text{O}$
  - CO<sub>2</sub> uptake can be estimated
  - Performance can be improved

# Benefits of Early Carbonation

- Technical:
  - Increased early age strength
  - Accelerated production
  - Reduced calcium hydroxide
- Environmental:
  - Permanent storage of CO<sub>2</sub> into a form of stable calcium carbonate
  - Carbon credit in a cap and trade system

## How Much CO<sub>2</sub> Can Be taken?

- The theoretical maximum of CO<sub>2</sub> uptake by Portland cement :
  - **$CO_2 \text{ (wt\%)} = 0.785 CaO + 1.09 MgO + 1.42 Na_2O + 0.935 K_2O$**
- At 100% carbonation, CaO is totally reacted with CO<sub>2</sub> to form CaCO<sub>3</sub> and carbon dioxide uptake can reach 50 wt%.

	CaO	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	Fe <sub>2</sub> O <sub>3</sub>	MgO
Portland cement	62.9	20.7	3.7	3.0	4.2

## Objectives

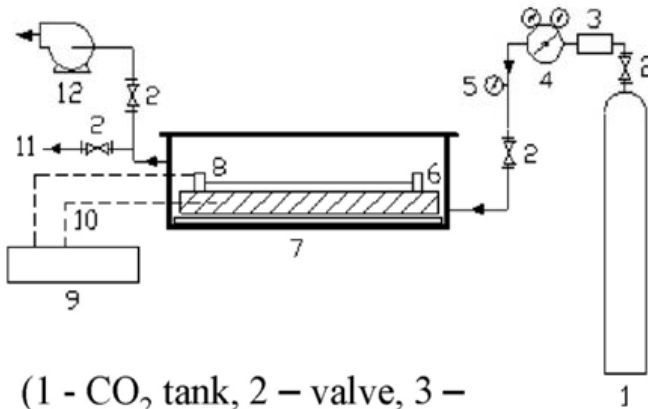
- To quantify carbon uptake capacities in dry mix concretes
  - Products: blocks, pavers, pipes, piles, hollow-core slabs, etc.
- To evaluate performance of carbonated products at different ages
- To perform cost analysis between carbon dioxide curing and steam curing

## Mixture Proportion

		<b>C</b>	<b>W/C</b>	<b>S/C</b>	<b>CA/C</b>	<b>Preset, hr</b>	<b>Carbonation time, hr</b>
<b>B1</b>	<b>Cement paste</b>	<b>1</b>	<b>0.15</b>	<b>0</b>	<b>0</b>	<b>0</b>	<b>2</b>
<b>B2</b>	<b>Cement paste</b>	<b>1</b>	<b>0.15</b>	<b>0</b>	<b>0</b>	<b>17</b>	<b>2</b>
<b>B3</b>	<b>Cement paste</b>	<b>1</b>	<b>0.15</b>	<b>0</b>	<b>0</b>	<b>0</b>	<b>18</b>
<b>B4</b>	<b>Cement paste</b>	<b>1</b>	<b>0.15</b>	<b>0</b>	<b>0</b>	<b>17</b>	<b>18</b>
<b>B5</b>	<b>Concrete</b>	<b>1</b>	<b>0.26</b>	<b>1.3</b>	<b>2.6</b>	<b>0</b>	<b>2</b>
<b>B6</b>	<b>Concrete</b>	<b>1</b>	<b>0.26</b>	<b>1.3</b>	<b>2.6</b>	<b>17</b>	<b>2</b>
<b>B7</b>	<b>Concrete</b>	<b>1</b>	<b>0.26</b>	<b>1.3</b>	<b>2.6</b>	<b>0</b>	<b>18</b>
<b>B8</b>	<b>Concrete</b>	<b>1</b>	<b>0.26</b>	<b>1.3</b>	<b>2.6</b>	<b>17</b>	<b>18</b>

- Early carbonation parameters:
  - Recovered CO<sub>2</sub> (99%) at gas pressure of 1.5 bar.
  - Sample size: 76x127x30 mm & 25x25x254 mm

# Carbon Uptake by Cement Binder



(1 - CO<sub>2</sub> tank, 2 – valve, 3 – heater, 4 – regulator, 5 - pressure gauge, 6 - bar sample, 7 - pressure vessel, 8 - LVDT assembly, 9 - data acquisition system, 10 – thermocouple, 11 – discharge, 12 - vacuum pump. )

1) Infrared (IR) based carbon analyzer method

$$CO_2 \text{ uptake}(\%) = \frac{(Mass)_{CO_2@800C}}{(Mass)_{dry \ binder}}$$

2) Mass gain method

$$CO_2 \text{ uptake}(\%) = \frac{(Mass)_{aft,CO_2} - (Mass)_{bef,CO_2}}{(Mass)_{dry \ binder}}$$



# Carbon Uptake

## Cement paste:

	T, °C (bar)	WL,% (bar)	Uptake %, (bar)	T, °C (slab)	WL,% (slab)	Uptake, %, (slab)	CO <sub>2</sub> content, % (slab)		
							surface	core	Ave.
B1	51.6	13.99	10.65	89.4	20.76	10.83	10.21	8.83	9.52
B2	46.6	5.71	9.58	77.0	11.87	10.85	9.52	8.95	9.24
B3	47.9	14.41	13.78	97.3	22.41	12.68	11.70	10.08	10.89
B4	41.4	3.19	13.29	86.2	13.73	13.22	13.46	12.04	12.75

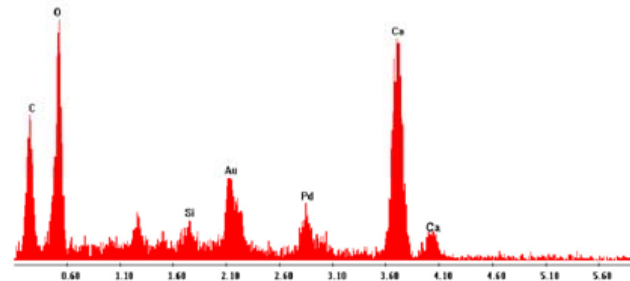
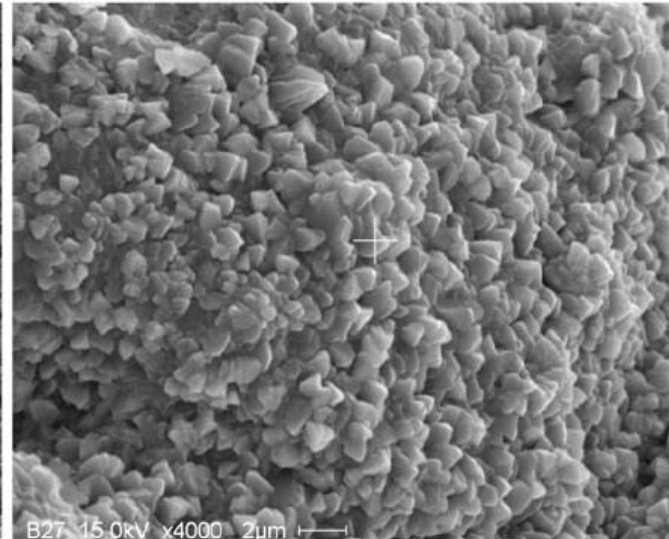
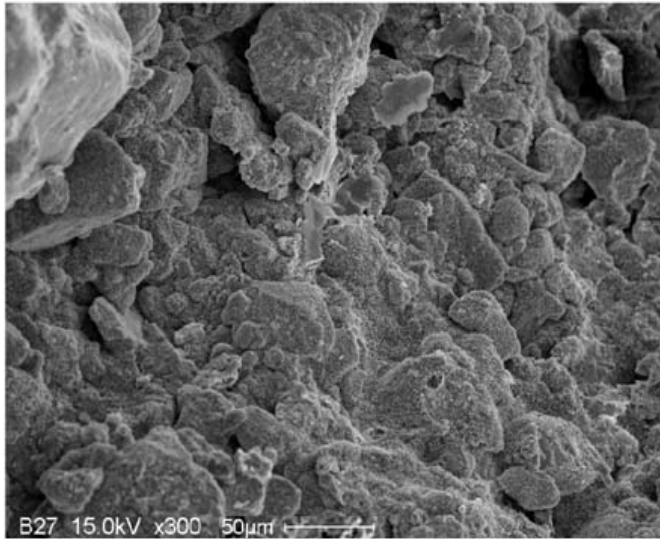
Note: T: peak temperature; WL: water loss; Uptake: CO<sub>2</sub> uptake based on Eq 1; CO<sub>2</sub> content: measured by CO<sub>2</sub> analyzer; Bar: measured from bar sample; Slab: measured from slab samples.

## Concrete

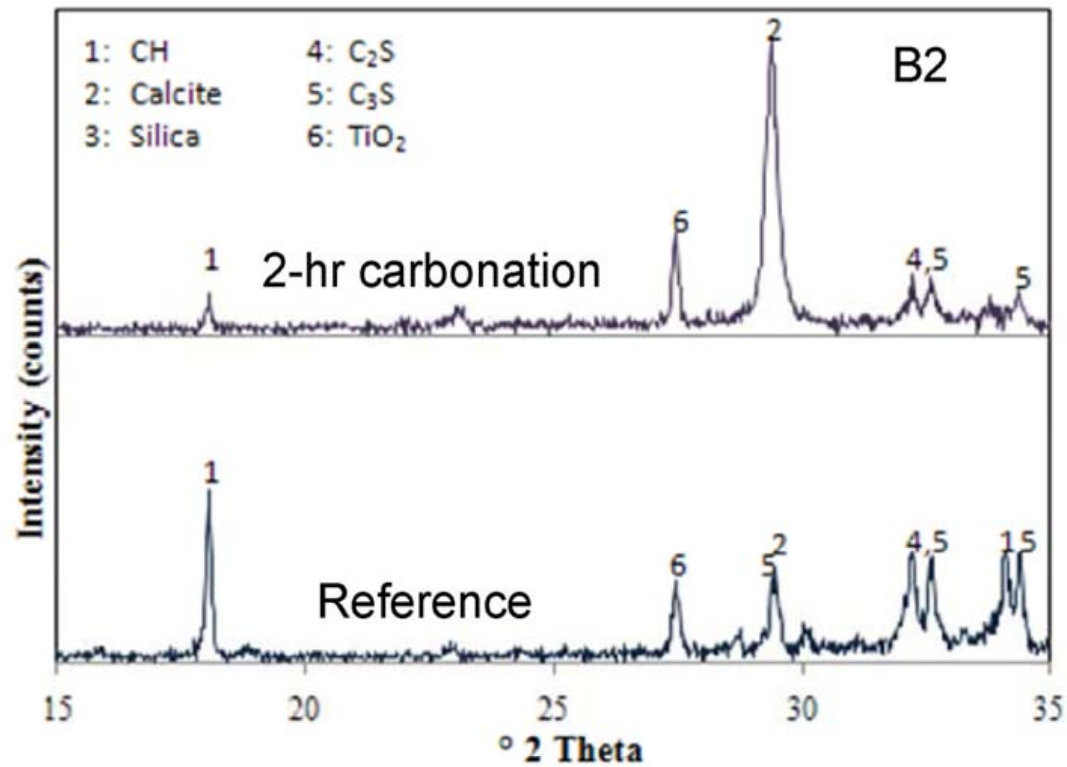
	T, °C (bar)	WL,% (bar)	Uptake, % (bar)	T, °C (slab)	WL, % (slab)	Uptake, % (slab)	CO <sub>2</sub> content, % (slab)		
							Carb.	Hyd.	Diff.
B5	40.2	7.88	9.70	59.9	16.49	10.15	15.22	7.22	8.0
B6	41.0	0.65	7.38	43.1	6.67	8.37	16.10	8.36	7.74
B7	41.4	6.31	12.98	56.9	17.99	15.02	20.39	7.52	12.87
B8	37.6	0.60	11.75	41.7	5.93	10.46	20.11	9.25	10.86

Note: T: peak temperature; WL: water loss; Uptake: CO<sub>2</sub> uptake based on Eq 1; CO<sub>2</sub> content: measured by CO<sub>2</sub> analyzer; Bar: measured from bar sample; Slab: measured from slab samples; Carb: carbonated; Hyd: hydrated; Diff: difference (CO<sub>2</sub> content due to carbonation).

# SEM of Carbonated Concrete



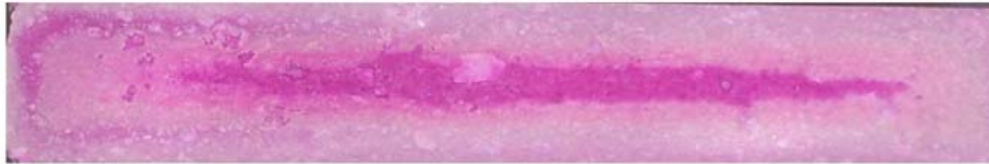
# XRD Patterns



After 7-day subsequent hydration

## Effect of Subsequent Hydration

Phenolphthalein tests of carbonated cement:



After 2 hours, uptake=13%

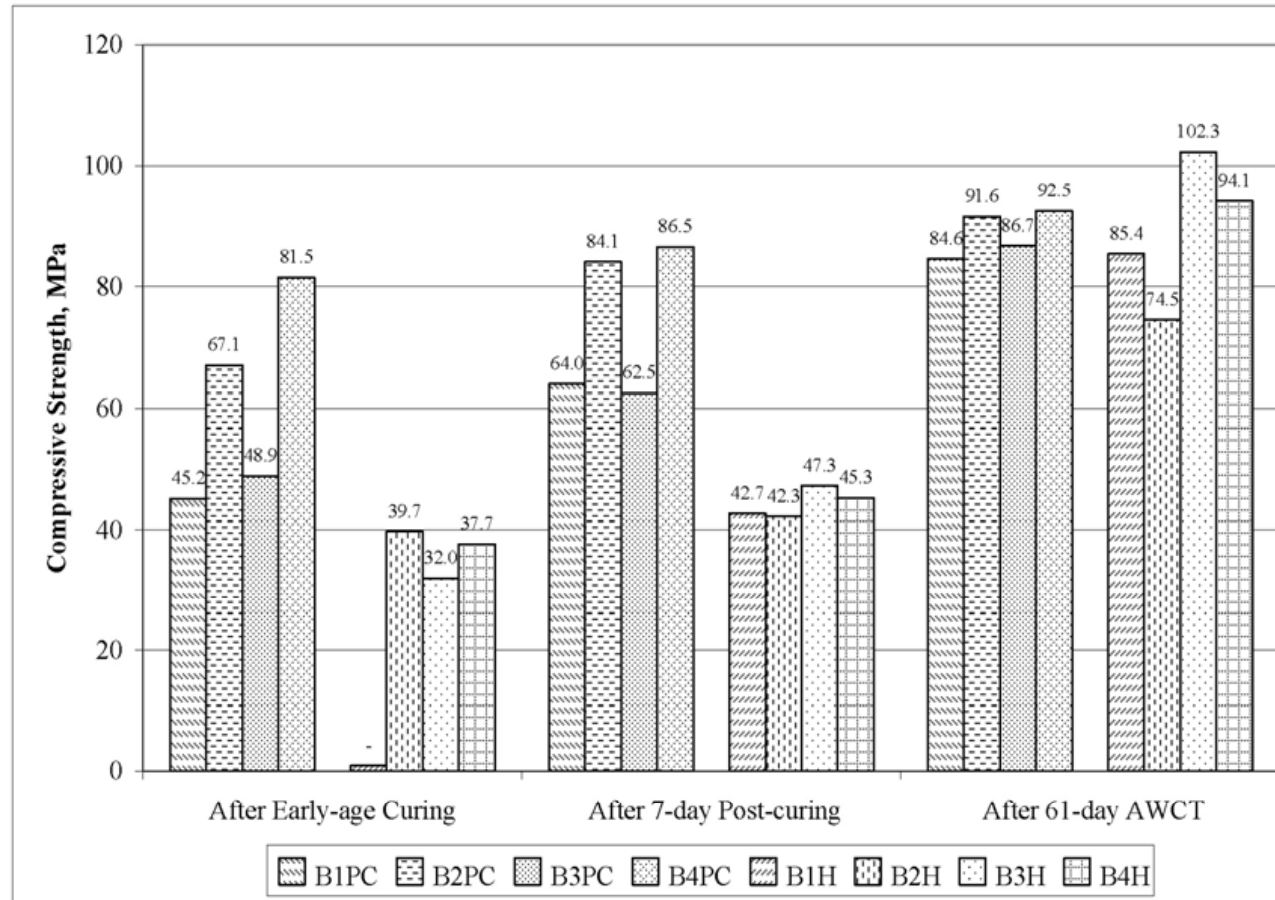


After 24 hours in sealed bag

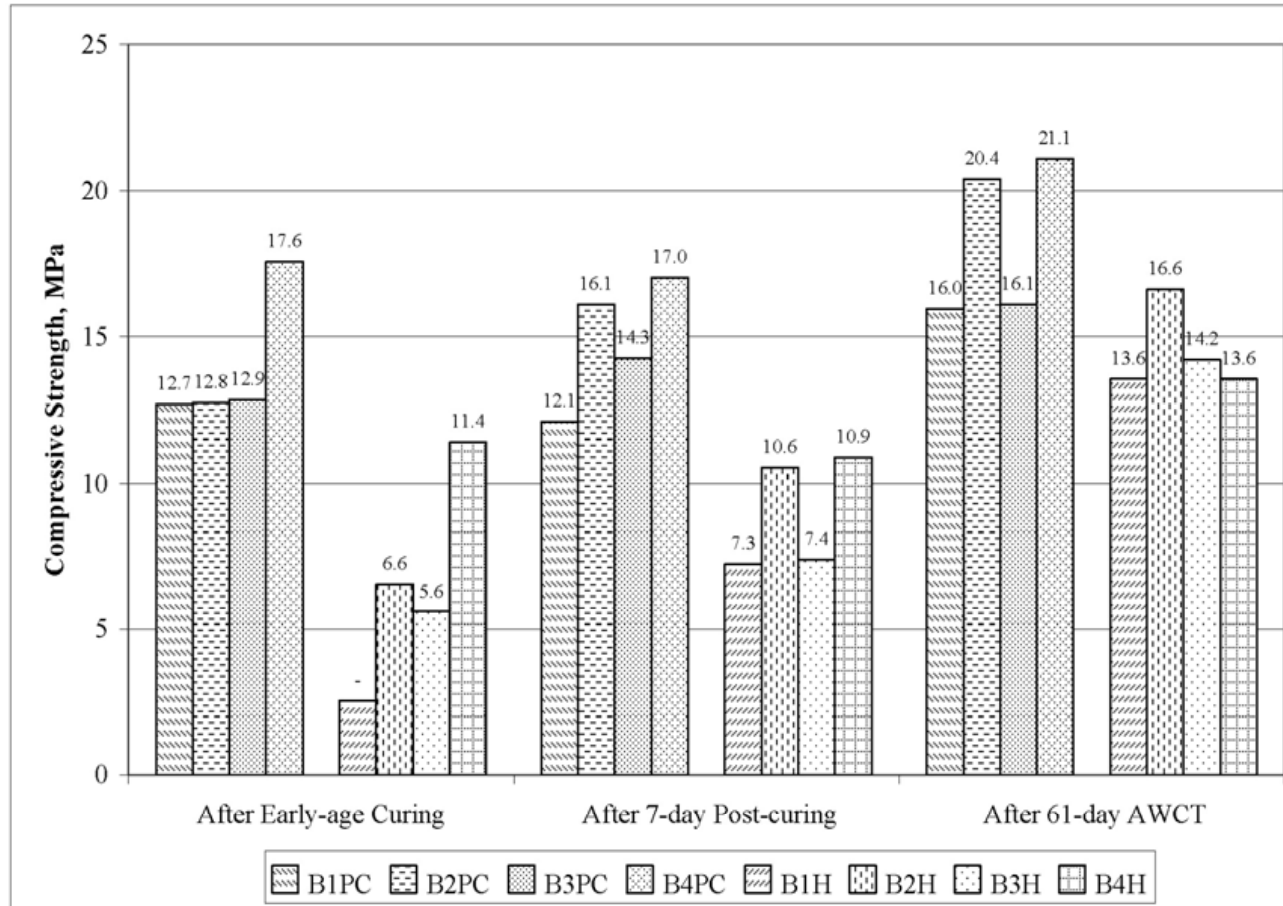


After 28 days in sealed bag

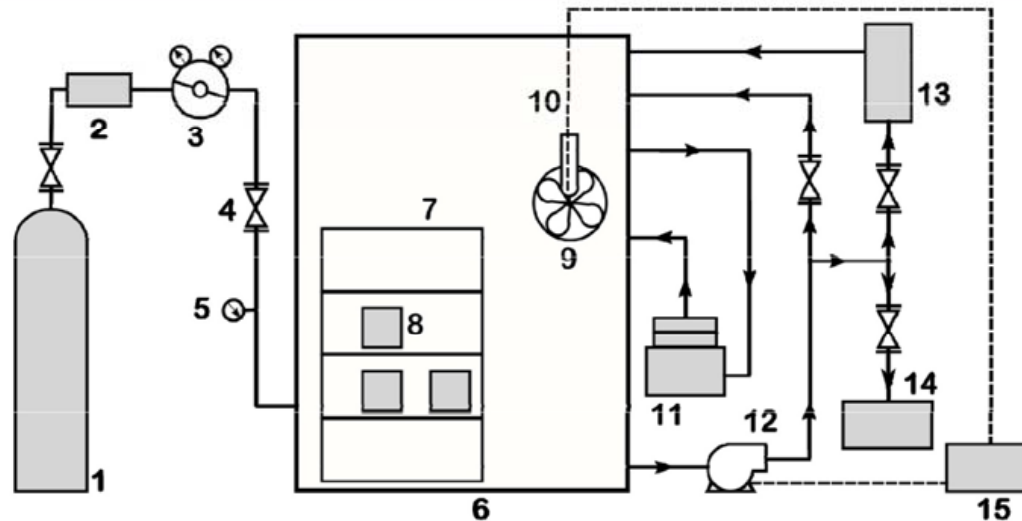
# Strength in Cement Paste



# Strength of Concrete



# Atmospheric Weathering Carbonation Test (AWCT)



1 - gas tank

2 - heater

3 - regulator

4 - valve

5 - pressure gauge

6 - AWCT chamber

7 - shelves

8 - samples

9 - fan

10 - humidity probe

11 - humidifier

12 - pump

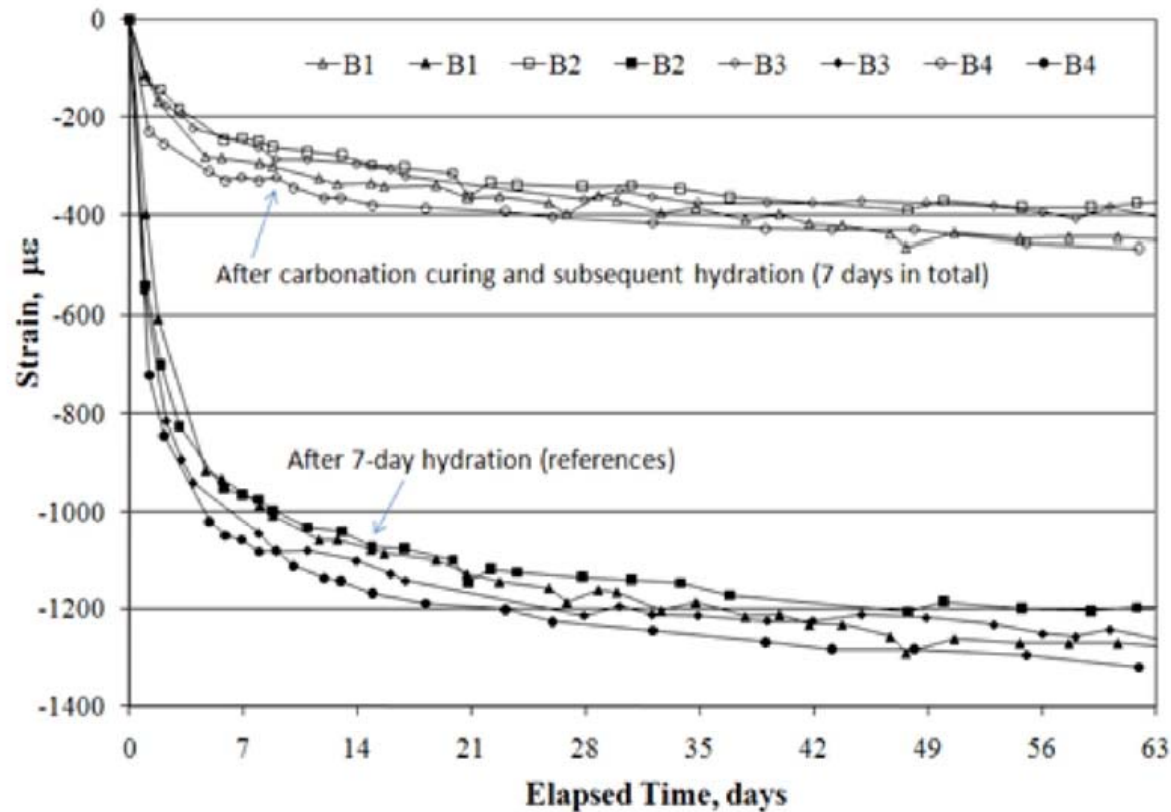
13 - desiccator

14 - CO<sub>2</sub> analyzer

15 - humidity controller

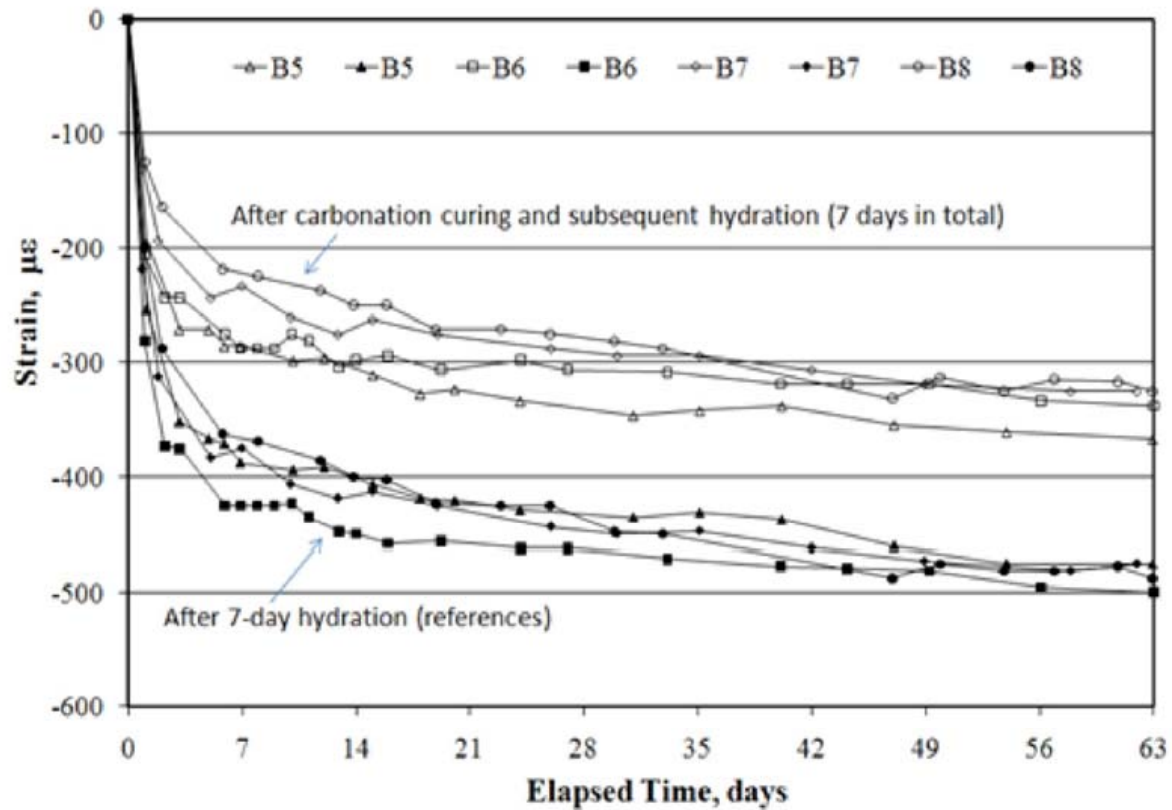
Condition: 50%CO<sub>2</sub>, 60% RH, 25 °C, 7-day curing.

# AWCT Shrinkage of Carbonated and Hydrated Cement





# AWCT Shrinkage of Carbonated and Hydrated Concrete



## Freeze-Thaw Resistance of Concrete

CSAA231.2-95 for 28-day concrete pavers:

- 16-hr freezing at -15 C in 3% sodium chloride solution
- 8-hr thawing at 22 C;
- Mass loss is calculated at 10, 25 and 50 cycles.

Sample	Carbonation Treatment	Cumulative Mass Loss, g/m <sup>2</sup>	
		10 Cycles	25 Cycles
SC	22 h Preset + 22 h CO <sub>2</sub>	218	1425
SH	Hydration	1762	10601



Carbonated, after 25 cycles



Hydrated, after 25 cycles

## Freeze-Thaw Resistance of Concrete Pavers

Sample	Carbonation Treatment	CO <sub>2</sub> uptake, %	24-hr Strength, MPa	Cumulative Mass Loss, g/m <sup>2</sup>	
				10 Cycles	25 Cycles
P1C	2 hr Preset + 4 hr	3.4	31	2.7	37.6
P2C	19 hr Preset + 5 hr	7.4	46	2.2	5.4
P3H	Hydration	0	35	26.9	328.0



Carbonated



Hydrated

# Cost Analysis

- Steam curing
  - Atmospheric steam curing:  $0.59 \text{ GJ/m}^3 = 164 \text{ kWh/m}^3$
  - Assuming  $\$0.1/\text{kWh}$ ,  $1\text{m}^3$  concrete costs  $\$16.4$ .
- Carbon dioxide curing
  - Assuming density =  $2400\text{kg/m}^3$ , cement content = 20%, and carbon uptake = 10% of cement
  - One cubic meter concrete needs 48 kg  $\text{CO}_2$
  - If cement producer captures  $\text{CO}_2$  at a price of  $\$50/\text{t}$  and sells it to concrete producer at  $\$80/\text{t}$ ,  $1\text{m}^3$  concrete costs  $\$3.84$ .

# Carbonation of Pervious Concrete

Mix Design	Unit	Unit/m3
Coarse Aggregate	kg	1498.00
Fine Aggregate	kg	104.00
Cement	kg	343.00
Silica Fume		
Water	kg	93.00
Air Entraining Agent	ml	
Super Plasticizer	ml	
W/C		0.2711
Paste/Aggregates		0.2722
Fine/Coarse		0.0694



Average  
carbon uptake  
in 2 hr = 8%

- Demoulded in 4hr;
- Dried in air overnight (19hr);
- Carbonated in a chamber for 2 hr at 1.5 bar with pure CO<sub>2</sub>;
- Tested at 24 hr, 7 days and 28 days

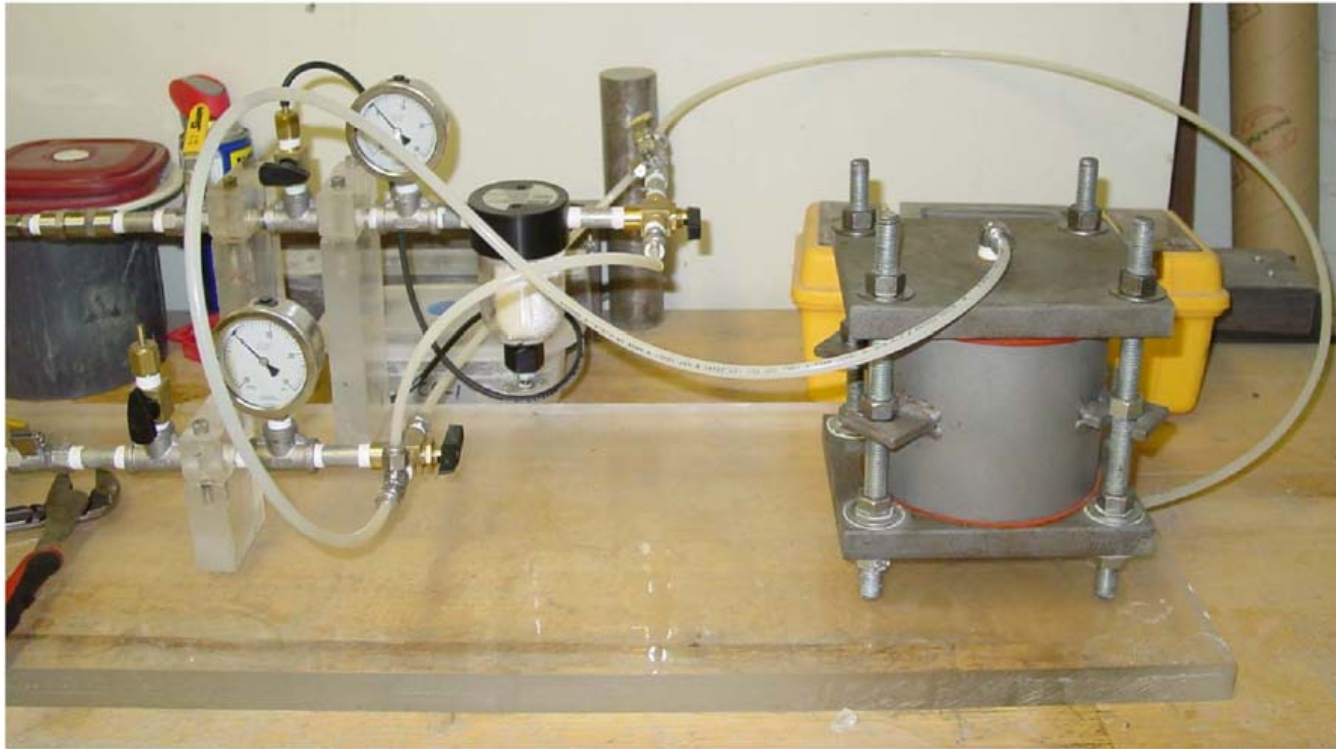
## Performance of Carbonated Pervious Concrete

Age (Days)	Reference (Non Carbonated) Compressive strength (Mpa)		Carbonated Samples Compressive strength (Mpa)	
	1	8.223 ±	0.869	12.251 ±
7	13.047 ±	2.259	11.817 ±	0.776
14	13.190 ±	2.403	11.397 ±	1.515
28	16.056 ±	1.845	16.115 ±	1.614

Age (Days)	Reference (Non Carbonated) k (cm/s)		Carbonated Samples k (cm/s)	
	1	0.949 ±	0.366	1.272 ±
7	1.105 ±	0.242	1.265 ±	0.421
14	0.537 ±	0.072	0.854 ±	0.559
28	0.727 ±	0.564	1.001 ±	0.236

Age (Days)	Reference (Non Carbonated) Absorption (%)		Carbonated Samples Absorption (%)	
	1	4.84% ±	0.11%	4.08% ±
7	4.31% ±	0.12%	3.74% ±	0.04%
14	4.24% ±	0.03%	3.73% ±	0.04%

## In-Situ Carbonation Simulation



## Conclusions

- Carbon dioxide can be beneficially utilized in concrete production.
- Early carbonation has no detrimental effect on late hydration strength.
- High pH value of early carbonated concrete can be maintained, while  $\text{Ca(OH)}_2$  is eliminated.
- Carbonation curing can cost less than steam curing due to energy reduction.



## Challenge

- CO<sub>2</sub> curing is best suited to concrete products that have large specific surface area and use low w/c ratio.
- For thick products, the reaction efficiency will be reduced.
- To promote CO<sub>2</sub> utilization in concrete, an innovative system is needed to provide incentives for both cement and concrete producers.

# Acknowledgment

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